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Anne C. Starley  
*Utah State University*

Gregory Wilson  
*Utah State University*

Lisa Montierth Phillipps  
*Utah State University*

JR Dennison  
*Utah State University*

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# Predictive Formula For Electron Range over a Large Span of Energies



Anne Starley, Gregory Wilson, Lisa Phillipps, and JR Dennison  
USU Materials Physics Group  
Utah State University, Logan, UT 84332-4414

## I. Introduction to Range

The range, commonly known as the penetration depth, describes the maximum distance electrons can travel through a material, given an initial incident energy, before they lose all of their kinetic energy and come to a rest.<sup>1,2</sup> The primary energy loss mechanism which causes the electron to lose its kinetic energy is due to inelastic collisions within material. Due to the probabilistic nature of this mechanism, the Continuous Slow Down Approximation (CSDA) is often employed to simplify the problem where the stopping power is taken as a constant.

This idea is illustrated by a Lichtenburg discharged tree pictured in Fig. 1. This “tree” demonstrates a situation where accelerated high energy electrons comes to rest and deposit charge at a given range in an insulating material.<sup>3</sup> The side view of the Lichtenburg tree displays the melted plastic caused by the energy of the deposited incident electrons at a uniform penetration depth. Here the stored charge is dissipated through a discharge.<sup>1</sup>



Fig. 1. Front (Left) and side (Right) views of a Lichtenburg discharge tree. The white line (Right) indicates the narrow distribution of deposited charge from a ~1 MeV electron beam at R=3 mm in a PMMA sample.

## II. Original Model

The previously developed model predicts the energy-dependent range,  $R(E)$ , as a function of incident electron energy for materials found in the NIST ESTAR database. In a continuous composite analytic approximation to the range with a *single fitting parameter*,  $N_V$ , spanning incident energies,  $E$ , from <10 eV to >10 MeV, the following functions (Eqs. 2, 3, and 4) describe the energy-dependent range,  $R(E)$ , in terms of  $N_V$  and material parameters mass density  $\rho_{NV}$ , effective atomic number and weight  $N_A$  and  $M_A$  and band gap  $E_{gap}$ .<sup>1</sup>

$$R(E; N_V) = \begin{cases} \left[ \frac{E}{E_m} \right] \frac{\lambda_{IMFP}(E_m)(1-\exp[-1])}{(1-\exp[-\frac{E}{E_m}])^2} & ; E < E_m \\ \left[ \frac{E}{E_m} \right] \frac{\lambda_{IMFP}(E)}{1-\exp[-\frac{E}{E_m}]} & ; E_m \leq E \leq E_{HI} \\ bE^n \left( 1 - \left[ 1 + \left[ \frac{E/N_V}{m_e c^2} \right] \right)^{-2} \right) & ; E > E_{HI} \end{cases}$$

Eq. (1). Range for low, medium, and high energy regimes.

$$E_m = 2.8[E_{gap}^2 + E_p^2]^{\frac{1}{2}}$$

Eq. (2). Mean energy lost per collision.

$$E_p = \hbar \left( \frac{N_V N_A \rho_m q_e^2}{m_e \epsilon_0 M_A} \right)^{\frac{1}{2}}$$

Eq. (3). Plasmon energy.

Manual fits to these range equations and optimum values of  $N_V$  were found using data for about 20 well-known elements and compounds with range data from the ESTAR database.<sup>2</sup> Fig. 2 shows several approximate fits to the range data.

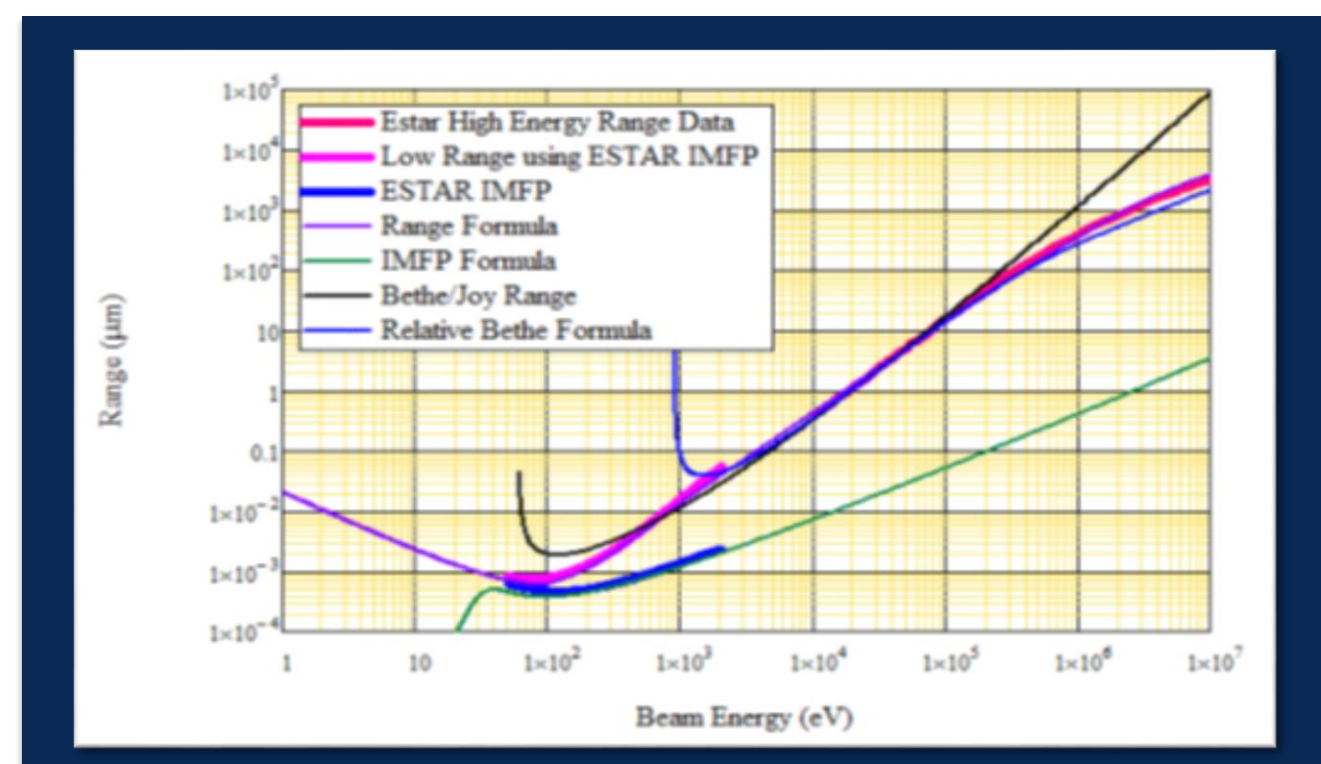


Fig. 2. Comparison between several range approximations and the data from the ESTAR database for Au. The IMFP data for Au are also plotted along with the TPP-2M IMFP formula for  $\lambda_{IMFP}(E)$ .

## III. Predictive Formula for $N_V$

A simple formula [Eq. (4)] was found to predict our single range parameter,  $N_V$ , as a function of only mean atomic number derived from the stoichiometric formula. This simple formula resulted from extensive analysis of much more complex predictive formula for  $N_V$  involving sums of power law terms for density, effective atomic number and weight, and bandgap plus other properties including plasmon energy, conductivity, phase, and more. This general fit for  $N_V$  was evaluated using general least squares fit analysis methods to simultaneously determine the best estimates for fitting parameters for each material property. Amazingly, the simple formula Eq. (4) involving only  $Z_m$  was the result.

Fig. 3 and Table 1 show the results of the fits with Eq. (4) for A,B and C. The fitting parameters was then used to calculate an estimate of  $N_V$  using the power law model. Plotting this estimate of  $N_V$  versus the true value of  $N_V$ , Goodness of fit metrics  $\chi^2_{NV}$  and R allows quantification of the quality of the fits.

To refine Eq. (4), separate fits were made for materials subcategorized into grouping such as insulators, semiconductors, and conductors [see Figs. 3(b-d)] and solids, liquids, and gasses with the hope that this categorization might unearth additional information. Semiconductor show excellent agreement. Insulators show very good agreement, with a slight downward concavity. Although conductors show good agreement, it is apparent that an additional higher order correction for conductors needed is needed to account for electron overlap in the d and f orbitals of transition and rare earth/actinide elements.

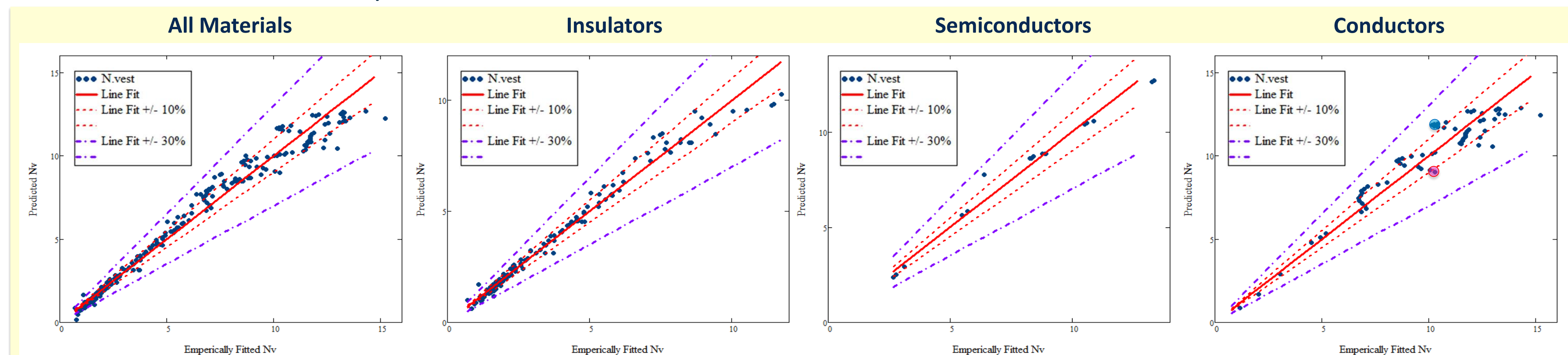


Fig. 3. The graphs examine  $N_V$  found from fits to the NIST database data versus  $N_V$  predicted using Eq. (4). Nominally, for exact agreement the slope of the fit would be 1 with an intercept of 0 and  $\chi^2_{NV}$  approaching 0. 10% and 30% error lines to the slope are marked in dashed red and dashed purple lines, respectively. Values for the constants for predictive  $N_V$  from Eq. (4) and the goodness of linear fit for Fig. (3) graphs are found in Table 1.

## IV. Accuracy of Range for Predicted $N_V$

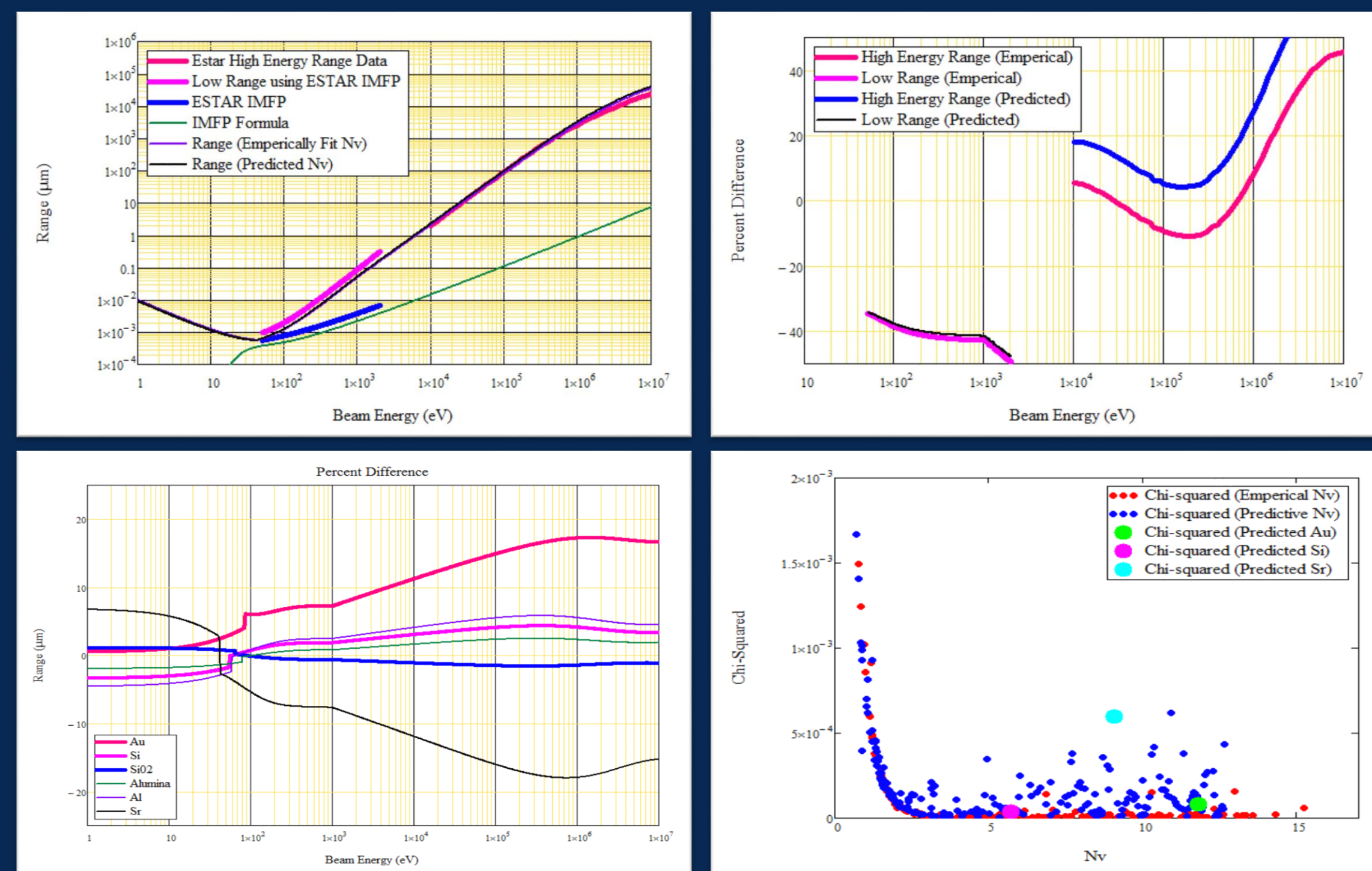


Fig. 4. (a) Strontium range calculated using both empirical and predicted  $N_V$  values as compared to ESTAR range data. (b) Percent differences for both medium and high energy regimes for Sr range calculated ranges for both empirical and predicted  $N_V$  values as compared to ESTAR range and inelastic mean free path data. (c) Percent differences between ranges calculated with empirical and predicted  $N_V$  values for  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ , Si, Al, Au and Sr. (d)  $\chi^2$  values for comparison of ESTAR range data to ranges calculated for both the empirical and predicted  $N_V$  values for all materials in the NIST material database (see Table 3). The  $\chi^2$  values for Si, Au and Sr with predicted  $N_V$  are highlighted, illustrating their extreme variances from the empirical data.

$$N_V(Z_m) = AZ_m^B - C$$

Eq. (4). Predictive formula for effective number of valence electrons, where  $Z_m$  the is mean atomic number and A, B, and C are constants found through fits shown in Fig. 3.

Table 1. Constants and Goodness of fit predictive  $N_V$  fits.

| Materials      | A      | B     | C      | $\chi^2_{NV}$ | R    |
|----------------|--------|-------|--------|---------------|------|
| All            | 20.196 | 0.653 | 21.727 | 0.006         | 0.99 |
| Insulators     | 21.581 | 0.730 | 22.892 | 0.0032        | 0.99 |
| Conductors     | 22.898 | 0.604 | 24.982 | 0.0053        | 0.99 |
| Semiconductors | 14.817 | 0.153 | 16.585 | 0.0055        | 0.99 |

## V. Materials with Predicted Ranges

| Materials | Elements | Organic Compounds |          | Polymer  |          | Composites |          |       |
|-----------|----------|-------------------|----------|----------|----------|------------|----------|-------|
|           |          | Formula           | Weight   | Formula  | Weight   | Formula    | Weight   |       |
| Metals    | Aluminum | 13.536            | Aluminum | 2.400    | Aluminum | 2.400      | Aluminum | 2.400 |
|           | Aluminum | 13.536            | Aluminum | 2.400    | Aluminum | 2.400      | Aluminum | 2.400 |
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